

# Thermoelectric properties of $\beta$ -FeSi<sub>2</sub> with Co and Ni addition

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$\beta$ -FeSi<sub>2</sub> crystallized in orthorhombic structure with  $Cmce$  space group is an abundant and eco-friendly material. In addition, this compound is promising in thermoelectric applications because of its good thermal stability and strong oxidation resistance at high temperatures<sup>1)</sup>. However, at high temperatures, the bipolar effect occurs due to a low carrier density ( $n_H$ ) with a narrow band gap of about 0.7 eV. The presence of the bipolar effect remarkably deteriorates the Seebeck coefficient ( $S$ ). As a result, the power factor ( $PF = S^2\rho^{-1}$ , where  $S$  is the Seebeck coefficient and  $\rho$  is electrical resistivity) also significantly decreases. It was reported that the addition of Co to  $\beta$ -FeSi<sub>2</sub> effectively increases the  $n_H$  and stabilizes  $S$  with temperature<sup>2)</sup>. In addition, the  $\rho$  decreases with Co doping, but it is independent of temperature which is probably due to small carrier mobility ( $\mu_H$ ). Therefore, in this work, we attempt to improve the  $\mu_H$  of Co-doped  $\beta$ -FeSi<sub>2</sub> by Ni addition. TE properties of  $\beta$ -Fe<sub>0.97-x</sub>Co<sub>0.03</sub>Ni<sub>x</sub>Si<sub>2</sub> ( $0 \leq x \leq 0.03$ ) are reported from 80 to 800 K. The bulk

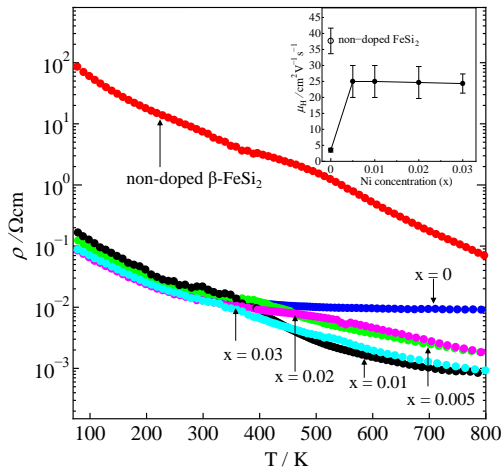


Fig. 1. Resistivity ( $\rho$ ) of pure  $\beta$ -FeSi<sub>2</sub> and  $\beta$ -Fe<sub>0.97-x</sub>Co<sub>0.03</sub>Ni<sub>x</sub>Si<sub>2</sub> ( $0 \leq x \leq 0.03$ ). The mobility ( $\mu_H$ ) is plotted as the inset.

samples were prepared by arc melting method and followed by heat treatment at 1150 °C for 3 hours and 840 °C for 20 hours. The X-ray diffraction data were measured by a CuK $\alpha$  diffractometer (SmartLab, Rigaku). The elemental analysis was performed by a scanning electron microscope (SU8010, Hitachi High-Technologies) equipped with an EDS XFlash5060FQ detector. The  $\rho$  and  $S$  were measured by using ResiTest8300 (TOYO Co.) and homemade apparatus. The thermal conductivity

( $\kappa_{total}$ ) was measured by using a power efficiency measurement (PEM-2, ULVAC, Inc.). Then, the  $ZT$  can be calculated by  $ZT = S^2T / (\rho\kappa_{total})$ . The result shows that the metallic  $\epsilon$ -phase increases with Ni and the solubility of Ni for  $\beta$ -phase is 1%.

The presence of the  $\epsilon$ -phase decreases the solubility of both Co and Ni in the  $\beta$ -phase. As Ni is added, the  $|S|$  decreases because of the formation of the  $\epsilon$ -phase. Fig. 1 shows that the  $\rho$  decrease with Ni doping because of the improved  $\mu_H$  as shown in the inset of Fig. 1. The  $\kappa_{total}$  slightly decreases with Ni probably due to the porosity. Consequently, the highest  $PF = 2400 \mu\text{Wm}^{-1}\text{K}^{-2}$  and  $ZT = 0.31$  is obtained in the 1% Ni-substituted sample.

## References

- 1) Y. Isoda and H. Uono, in *Thermoelectrics and its Energy Harvesting*, ed. D. M. Rowe (CRC Press, 2012) 1st ed. pp. 18-1.
- 2) S. Sam, H. Nakatsugawa and Y. Okamoto, *Jpn. J. Appl. Phys.* **61**, 111002 (2022).